

Modelling tritium in aquatic environment

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Why are we interested in dynamic models for the dose assessment of liquid releases ?

- 1. Some processes cannot be described by steady-state models : discontinuous process such as sediment deposit and resuspension
- Steady state models, used to demonstrate compliance with regulatory dose limits, are difficult to validate in the environment where concentrations change according to time in the day, season, river discharge,...Case of NPP liquid releases, discontinuous process and time-dependent pathways (irrigation) Validation is possible by :
 - Comparing dynamic models to field data
 - Running dynamic models on a longer time range (year) and comparing yearly average results with steady state model to check that they are conservative
- Dynamic models useful to demonstrate that different turn-over rates for HTO and OBT can explain observed OBT/HTO >1









From discharge in water to man : review of existing tritium models

Oispersion/transport in river or sea

Transfer to aquatic organisms

 Transfer through irrigation to agricultural products



Review of aquatic tritium models

- Literature review in 2000 : Steady state specific activity models with or without OBT
- In 2000, CALVADOS (later called OURSON) dynamic model for tritium and carbon 14 in aquatic environment applied on the Loire river
- In 2004, IAEA EMRAS intercomparison exercises

odispersion /transport model ____ Loire river scenario

odynamic transfer to mussel transplantation scenario in Perch lake



ono scenario with irrigation

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EMRAS - A comprehensive Loire river basin scenario from 1994 to 1999





EMRAS scenario on tritium migration in the Loire river

- Simulation of the dispersion of Tritium discharges in the whole Loire river system (~ 350 km)
- Reproduction of the real hydraulic conditions, from July to December 1999

- Comparison between calculated tritium concentration and measurements at Angers
- Inter-comparison between the different models at different points along the river



Models tested on the Loire scenario

GOUTAL et al., 2008, Journal of Environmental Radioactivity

CASTEAUR

IRSN, France

- MASCARET TRACER module
- MOIRA+ MARTE module

ENEA, Italy

EDF. France

RIVTOX

IMMSP, Ukraine

Conclusions :

•Good agreement between model and measurements for <u>average</u> concentrations

 Performance of models controlled by appropriate estimates of water velocities and water fluxes : 1D hydrological models better adapted to sharp release or high hydraulic variability





Models tested on the mussel transplantation scenario

- Model from NIRS, Japan
- Model from SRA, Japan
- OURSON EDF, France
- AQUATRIT IFIN, Romania
- BIOCHEM TUM, Germany

Conclusions :

Underestimation of OBT concentration in the first 24 h,overprediction after 88 days

•Understanding of processes involved in OBT dynamics need to be improved



General model for transfer to biota

HTO

• Rapid equilibrium between HTO in the organism and HTO in the surrounding media (air or water)

• Turn-over rate controlled by ratio between water intake and body water content

• OBT

• same general equation for OBT and carbon 14 in phytoplancton, fish, terrestrial plants and animals based on food intake rate or CO2 assimilation rate for photosynthetic organisms (Sheppard et al 2006)

$$\frac{d(A_{biota}^{C14} \cdot M_{biota}^{mass})}{dt} = A_{biota}^{C14}(t) \cdot \frac{dM_{biota}^{mass}(t)}{dt} + M_{biota}^{mass}(t) \cdot \frac{dA_{biota}^{C14}(t)}{dt} = A_{biota}^{C14}(t) \cdot M_{biota}^{mass}(t) + I \cdot K \cdot D \cdot A_{substrate}^{C14}(t) \cdot M_{biota}^{mass}(t) = A_{biota}^{C14}(t) \cdot M_{biota}^{mass}(t) + I \cdot K \cdot D \cdot A_{substrate}^{C14}(t) \cdot M_{biota}^{mass}(t)$$



Determination of turn-over rate of tritium in phytoplancton from phytoplancton growth model

•Phytoplancton growth model developped for eutrophication problem to predict O_2 evolution in aquatic environment :

- Growth is a function of light (photosynthesis), water temperature and nutrient availability
- Disappearance (respiration and predation)



ROD

Phytoplancton model





Questions to be addressed

- For living organisms, single compartment model has the advantage of simplicity, two compartment model with a fast and slow turn-over rate might be more accurate but parameter values difficult to establish
- Check the availability of food intake rate for different categories of aquatic organisms (molluscs, crustaceans, fish)
- Process to be included or not:
 - transfer from dissolved OBT(radiolabelled biomolecules) to aquatic organisms (done in AQUATRIT)
 - Transfer from sediment organic matter to bottom feeder requires to determine the bioavailability of OBT

• Transfer between atmosphere and water (done in MASCARET)



Compartments and pathways (including irrigation)for exposure to tritium from liquid releases (OURSON)

HTO riv	Éch. H ₂ O	Incorp.	Irrigation						Ingestion	Transformation	Ingestion
Éch. H ₂ O	HTO pois.										Ingestion
Elimin .bio		OBT pois.									Ingestion
			Sol - HTO	Infiltration	Evaporation	Prélèv. racinaire					
				Sol profond Nappe							
					Air						
					Transpir.	HTO – Feuil.	Photo- synthèse	Transloc.	Ingestion		Ingestion
							OBT – Feuil.			Assimilation	Ingestion
								OBT – grains	Ingestion	Assimilation	Ingestion
									HTO – Animal Part. com.		Ingestion
										OBT – Animal Part. com.	Ingestion
											Homme
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Uncertainty of mean annual dose (Sv/an) in Montjean (OURSON results)

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Fifth percentile	95th percentile	Ratio 95th percentile/fifth percentile
9.9×10^{-9}	4.1×10^{-8}	4.1
1.3×10^{-9}	4.4×10^{-9}	3.4
8×10^{-10}	2.7×10^{-9}	3.4
6.6×10^{-9}	1.7×10^{-8}	2.6
1.9×10^{-9}	$8.8 imes 10^{-9}$	4.6
	Fifth percentile 9.9×10^{-9} 1.3×10^{-9} 8×10^{-10} 6.6×10^{-9} 1.9×10^{-9}	Fifth percentile95th percentile 9.9×10^{-9} 4.1×10^{-8} 1.3×10^{-9} 4.4×10^{-9} 8×10^{-10} 2.7×10^{-9} 6.6×10^{-9} 1.7×10^{-8} 1.9×10^{-9} 8.8×10^{-9}

Source : Ciffroy ,Siclet et al , 2006, Journal of Environmental Radioactivity



Sensitivity analysis for ingestion of milk and ingestion of root vegetables

Ose due to ingestion of milk – sensitivity index



Source : Ciffroy ,Siclet et al , 2006, Journal of Environmental Radioactivity



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